

FLOW ELECTROCHEMISTRY FOR THE METAL ENERGY CYCLE

Prof. Dr.-Ing. John Linkhorst – Process Engineering of Electrochemical Systems

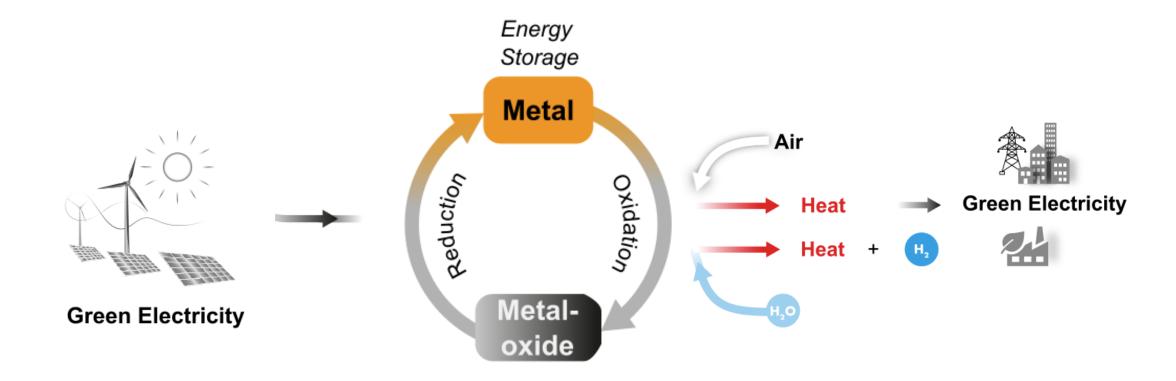
ERCOFTAC Autumn Festival 2025

9.10.2025

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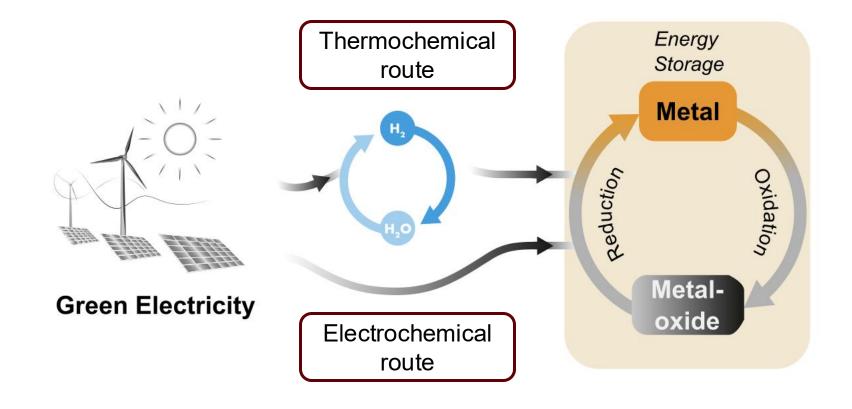
THE METAL ENERGY CYCLE



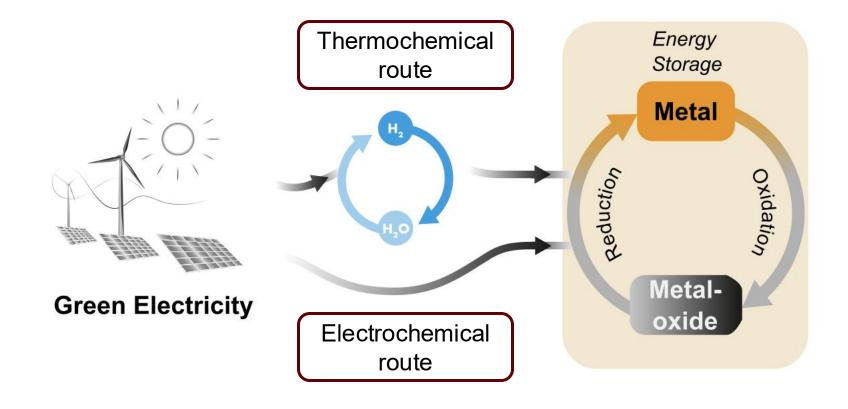


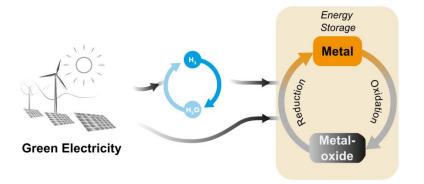
REDUCTION OPTIONS



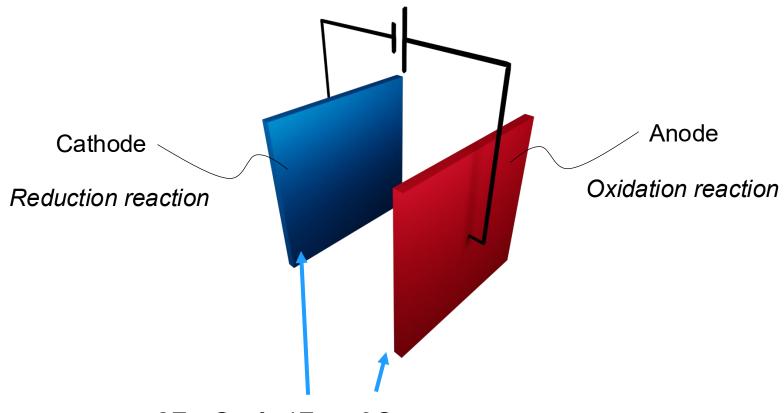




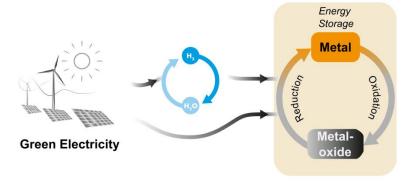




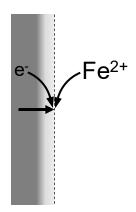




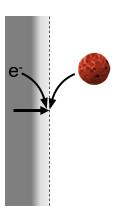
 $2Fe_2O_3 \rightarrow 4Fe + 3O_2$



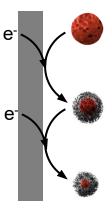




Acidic electrodeposition



Alkaline electrodeposition



Direct alkaline electroreduction



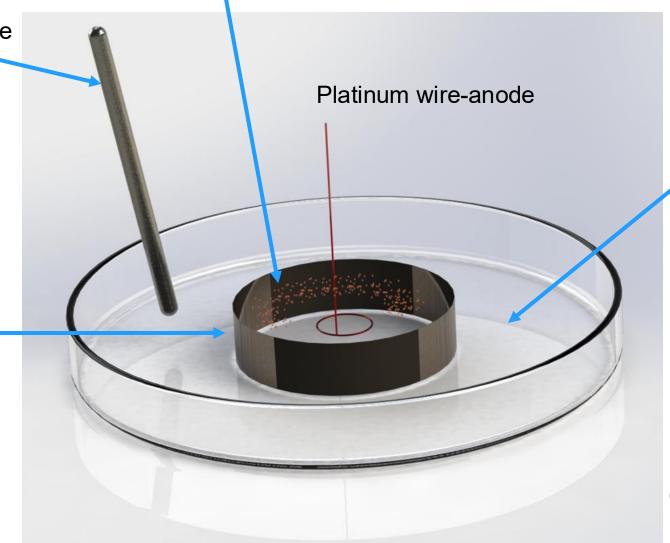
FEASIBILITY CHECK





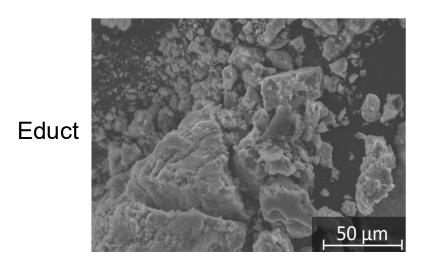
Temperature sensor

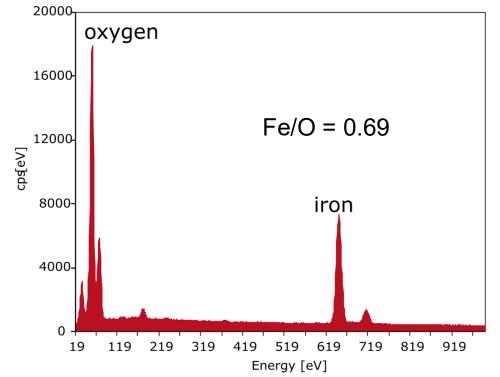
Nickel cathode

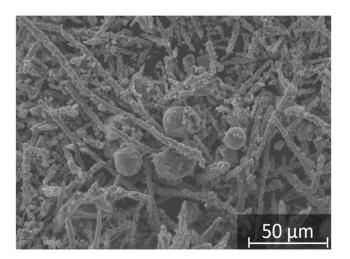


50 wt% NaOH in H₂O

Conditions: 110°C, 1.61V

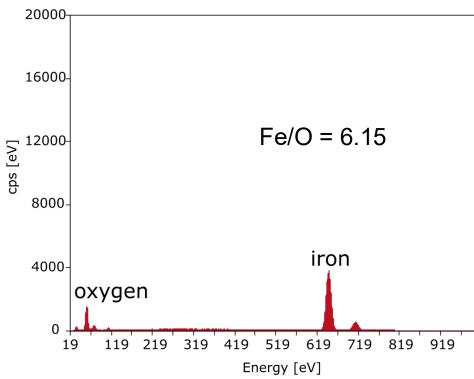








Product

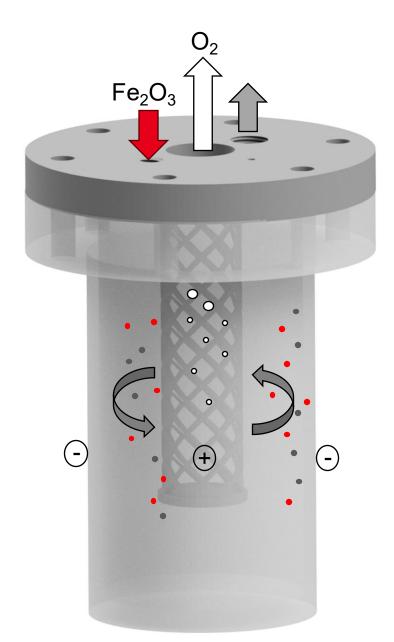




FLOW ELECTROLYSIS

CONTINUOUS SLURRY REACTOR



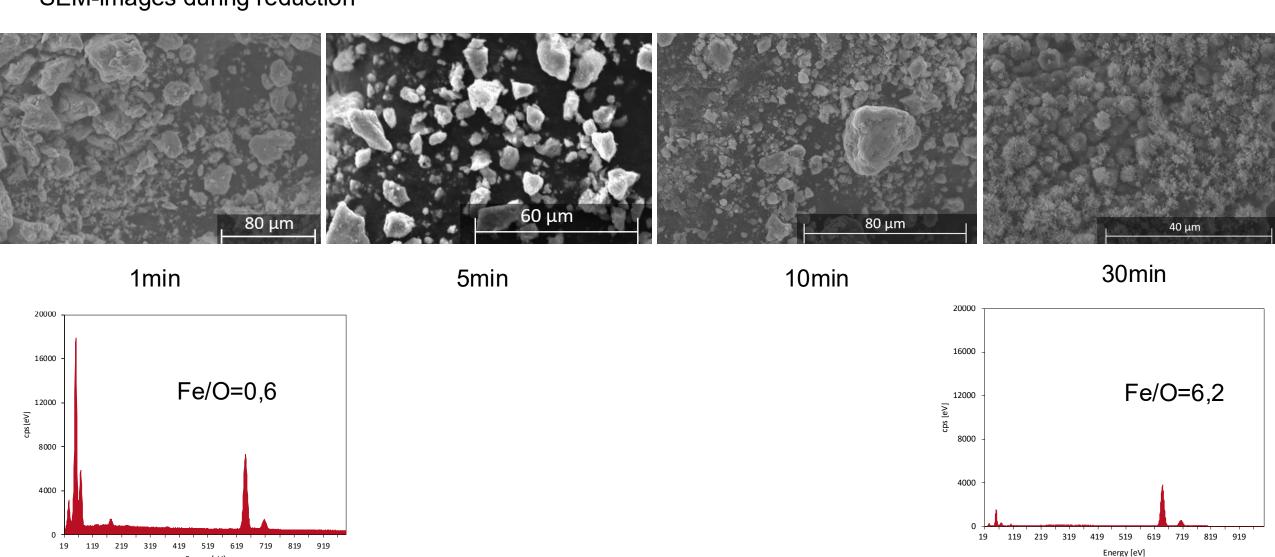


CONTINUOUS SLURRY REACTOR



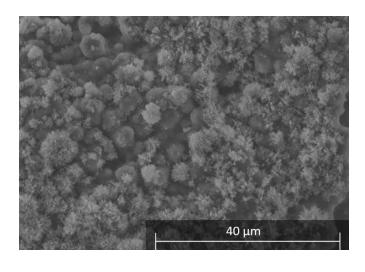
SEM-images during reduction

Energy [eV]

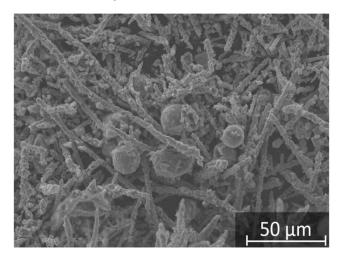


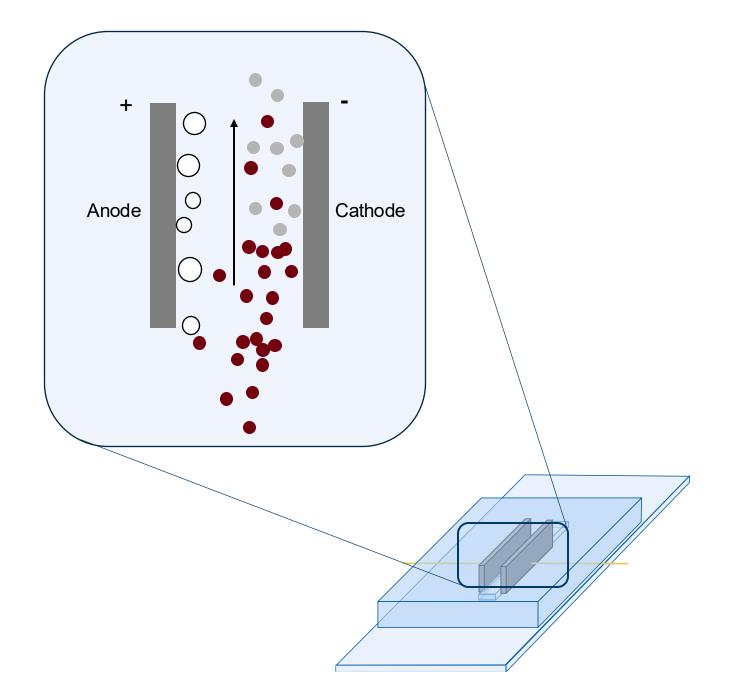


Flow reactor



Feasibility test

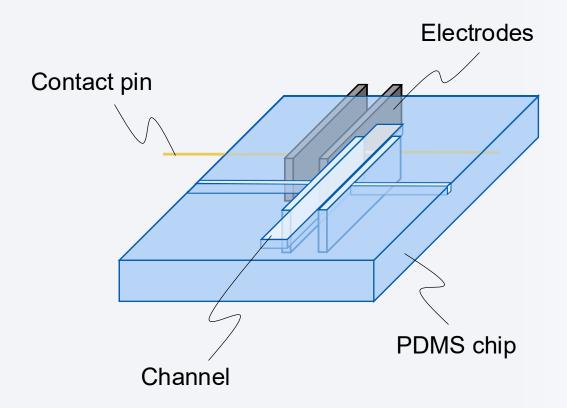


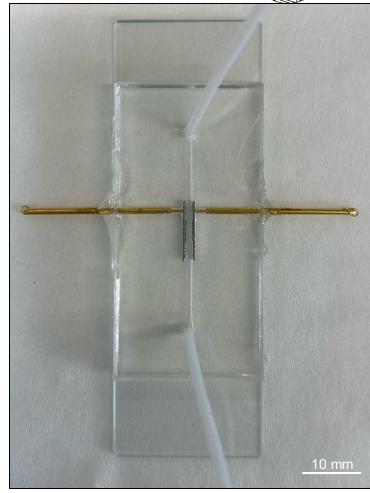




How do the particles interact with the electrode?

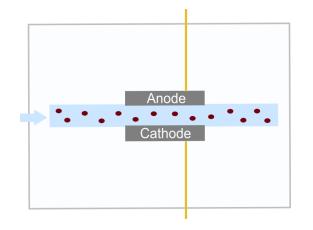


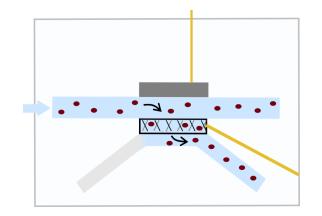


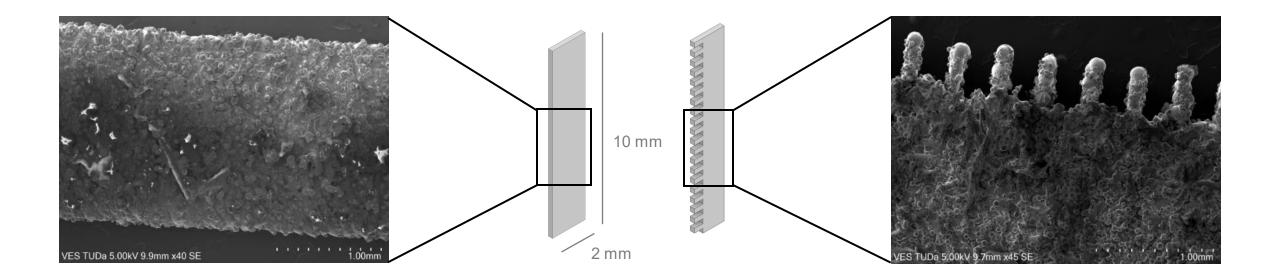


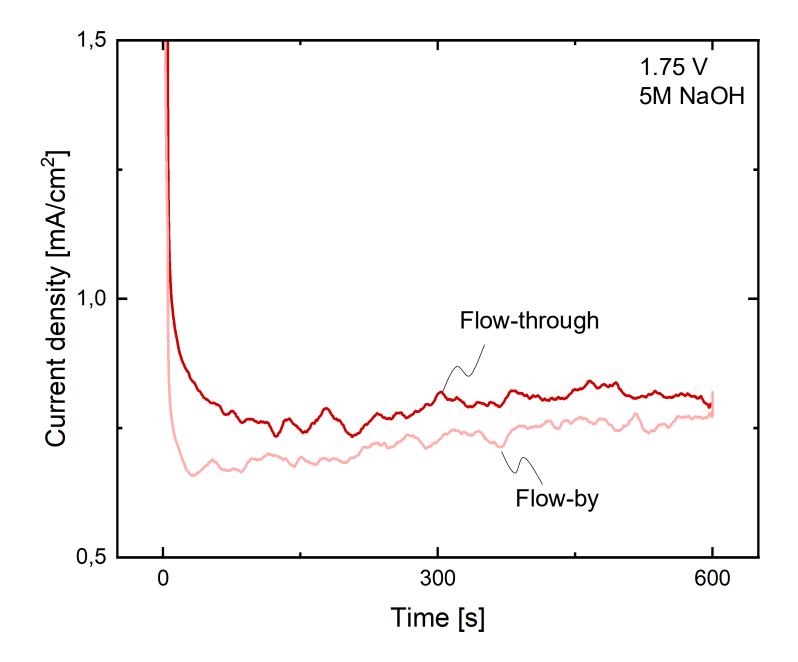
Flow-by

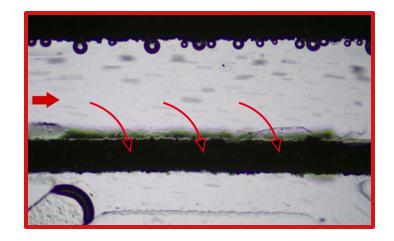
Flow-through

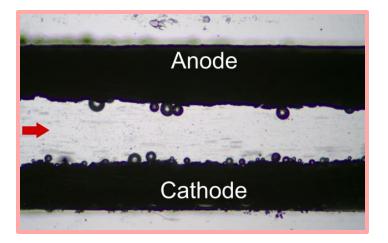






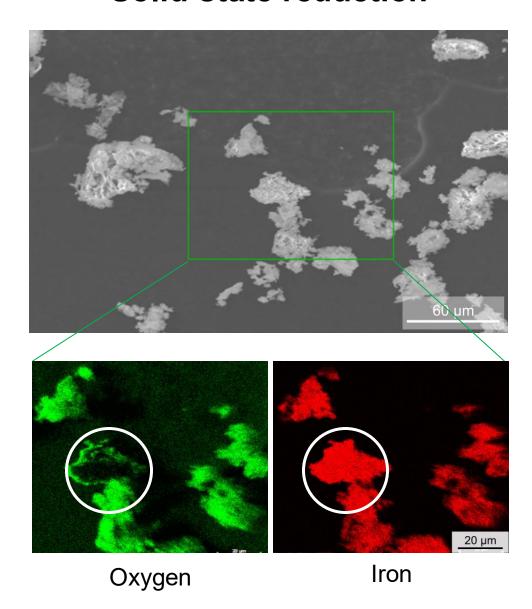






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Solid-state reduction



STAY TUNED FOR THE NEXT DEVELOPMENTS

